

## NOVEL CHIRAL PORPHYRINS WITH $C_2$ SYMMETRY<sup>1</sup>

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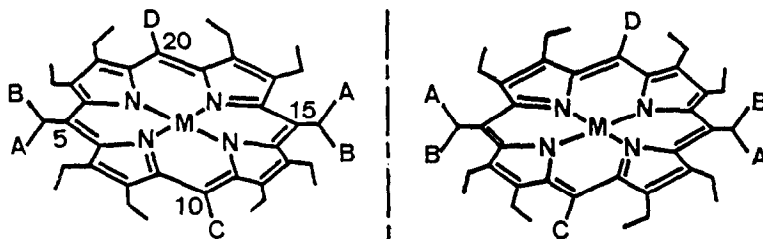
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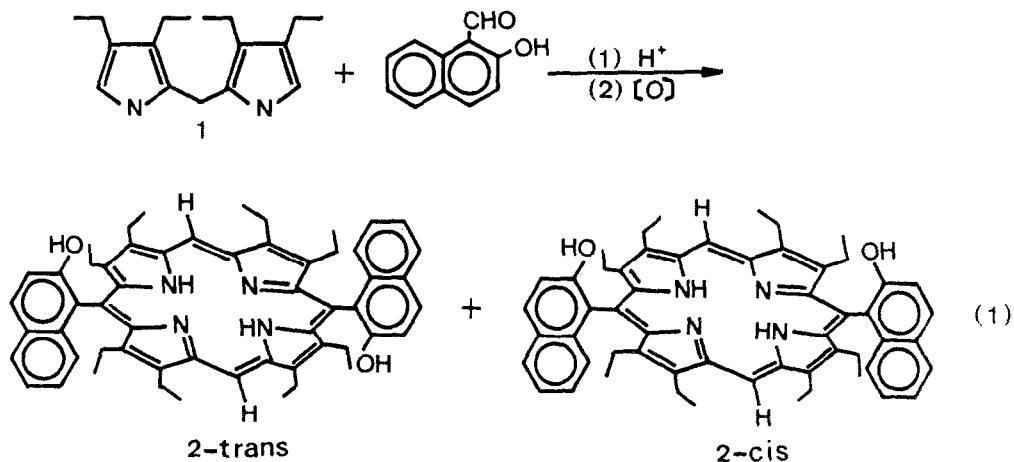
**Summary:** Described is the preparation of chiral octaethylporphyrin deriva-  
tives having two 2-substituted naphthyl groups in a trans orientation at the  
5- and 15-meso positions and formyl or hydroxymethyl or alkoxyethyl group  
derived therefrom at the 10-meso position. One of the derivatives has been  
successfully resolved into enantiomers by means of HPLC.

Metalloporphyrins are able to catalyze various types of organic reac-  
tions. Chiral modification of metalloporphyrins, therefore, is an interest-  
ing approach to the catalytic asymmetric organic synthesis.<sup>3</sup> In order to  
achieve a high degree of chiral recognition, the conformational flexibility of  
chiral porphyrin catalyst should be minimized and porphyrin-substrate inter-  
action should be maximized via multi-site interaction. We wish to report the  
preparation and chromatographic optical resolution of novel mesosubstituted  
chiral octaethylporphyrins, of the general structure shown below, with  $C_2$   
symmetry. The asymmetric spatial arrangement of meso-substituents (A, B, C,  
and D) is identical at both sides of the porphyrin plane, and is expected to  
provide a steady chiral space or reaction field surrounding the central metal,  
at which or in the vicinity of which reactions of substrates occur.



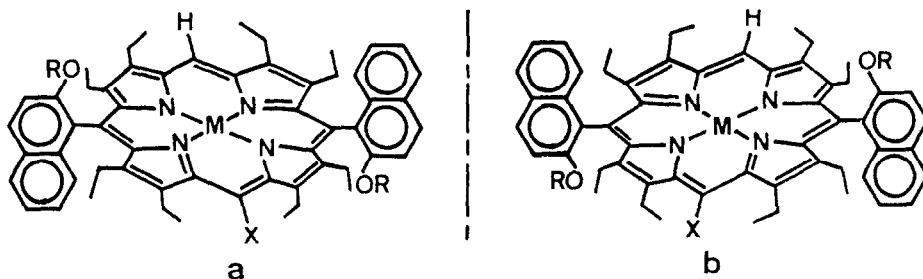
The reaction of 3,3',4,4'-tetraethyl-2,2'-dipyrromethane (1)<sup>4</sup> with 2-  
hydroxynaphthaldehyde by the procedure of either Ogoshi, *et. al.*,<sup>5</sup> (conden-  
sation in benzene with  $CF_3CO_2H$  followed by oxidation with  $O_2/2,6$ -lutidine/alumi-  
na) or Chang, *et. al.*,<sup>6</sup> (condensation in methanol with  $TsOH$  to give the  
porphyrinogen followed by its oxidation with chloranil in THF) for the prepa-  
ration of 5,15-diaryloctaethylporphyrin afforded 5,15-bis(2-hydroxynaphthyl)-

octaethylporphyrin (2-trans and 2-cis, eq 1). The atropisomers were separated by means of chromatography on silica gel with chloroform as an eluent into 2-trans (37%, TLC  $R_f$  0.67 (silica/ $\text{CH}_2\text{Cl}_2$ )) and 2-cis (27%, TLC  $R_f$  0.07), which showed almost identical spectroscopic properties.<sup>7</sup> The structures of atrop-



isomers were elucidated by intramolecular bridging of two *cis* hydroxyl groups; 2-cis reacted with 1,8-dibromooctane under basic conditions to give cyclophane porphyrin (3, 50%).<sup>8</sup> The rotation of naphthyl ring was found to be completely inhibited; there was no evidence for the interconversion of 2-trans  $\rightleftharpoons$  2-cis when a toluene solution of 2-trans or 2-cis was heated at 100°C for 6 h.

In order to convert 2-trans into a chiral porphyrin, various attempts were made to introduce one formyl group at the free meso-positions. The best results were obtained when 2-trans was first converted to its nickel complex (nickel acetate, 100%) and then formylated under Vilsmeier conditions ( $\text{POCl}_3/\text{DMF}$ )<sup>9</sup> to give racemic Ni(II) complex of 10-formyl-5,15-*trans*-bis(2-hydroxy-naphthyl)octaethylporphyrin (4a and 4b, 65%):  $\delta$  11.28 (s, 1H, CHO), 9.00 (s, 1H, meso);  $\nu_{\text{CO}}$  1620  $\text{cm}^{-1}$ ;  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 442 (4.99), 554 (3.69), 671 nm (4.09)



4 : R=H, X=CHO, M=Ni(II)

5 : R=H, X=CHO, M=2H

6 : R=H, X=CH<sub>2</sub>OH, M=Ni(II)

7 : R=CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>, X=CH<sub>2</sub>OCH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>, M=Ni(II)

for a  $\text{CH}_2\text{Cl}_2$  solution unless otherwise stated. Demetalation of 4 under acidic conditions ( $\text{HCl}/\text{CF}_3\text{CO}_2\text{H}$ ) afforded racemic free bases (5a and 5b,  $\sim 100\%$ ):  $\delta$  12.04 (s, 1H, CHO), 9.62 (s, 1H, meso);  $\nu_{\text{CO}}$  1620  $\text{cm}^{-1}$ ;  $m/e$  818 ( $\text{M}^+-\text{CO}$ ). This compound in  $\text{CH}_2\text{Cl}_2$  solutions underwent ready decarbonylation in a day at  $30^\circ\text{C}$  and in a week even at  $-20^\circ\text{C}$ . Reduction of the formyl group in 4 with  $\text{NaBH}_4$  suspension in ether gave the 10-hydroxymethyl derivative (6,  $\sim 100\%$ ):  $\delta$  6.03 (double d, 2H,  $\text{CH}_2\text{OH}$ );  $m/e$  905 ( $\text{M}^+$ );  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 429 (5.25), 554 (4.10), 592 nm (4.21). Alkylation of the three OH groups in 6 with benzyl bromide ( $\text{NaH}/\text{dioxane}$ ) afforded the tri(benzyl ether) derivative (7, 23%):  $\delta$  5.64 (s, 2H,  $10-\text{CH}_2\text{OCH}_2\text{C}_6\text{H}_5$ ), 5.07 (s, 4H,  $\text{OCH}_2\text{C}_6\text{H}_5$ ), *ca.* 3.45 (2H,  $10-\text{CH}_2\text{OCH}_2\text{C}_6\text{H}_5$  overlapping with  $\text{CH}_2\text{CH}_3$ );  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 235 (5.02), 283 (4.33), 324 (4.11), 428 (4.97), 552 (3.87), 592 nm (4.04) for a n-hexane-2-propanol (99:1) solution.

By means of HPLC using an optically active column prepared from cellulose tris(3,5-dimethylphenylcarbamate),<sup>10</sup> the racemic mixture of 7 was successfully resolved into two components (1:1) having retention times of 25 and 32 min.<sup>11</sup> These were unambiguously identified as enantiomers since they showed the CD spectra with opposite signs (Figure 1).

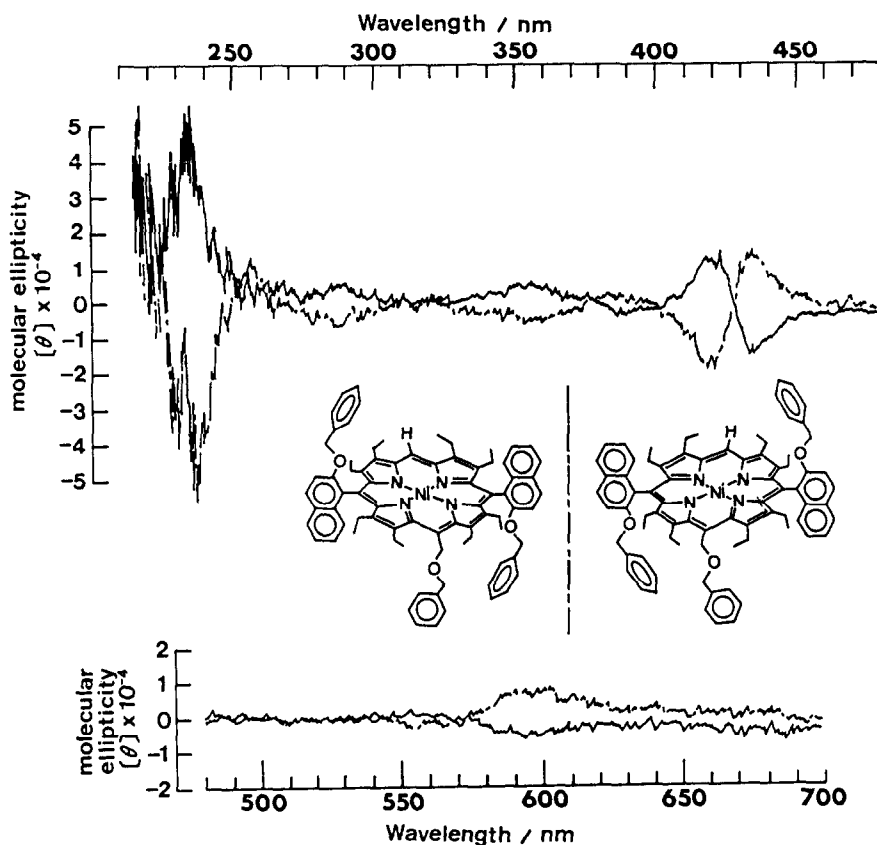


Figure 1. Circular dichroism (CD) spectra of optically pure enantiomers of 7 in n-hexane-2-propanol (99:1): enantiomer having retention time of 25 min (—) and that of 32 min (---). The spectra were obtained with a JASCO-J40 spectrometer using a cell of 1-cm path length,

Further work is now under way to obtain various derivatives, to establish the relationships between steric and electronic properties of substituents R and X and extents of chiral recognition, and to determine the absolute configuration of enantiomer by X-ray crystallography.

Acknowledgement. This work was supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Science, and Culture. We are grateful to Dr. Tamio Ueno (Kyoto University) for measurement of mass spectra and also to Mr. Hideaki Sakamoto (Osaka University) for carrying out HPLC optical resolution.

#### REFERENCES AND NOTES

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4. Prepared in 80% yield from 2-methyl-3,4-diethyl-5-benzyloxycarbonylpyrrole by successive reactions of oxidative dimerization [Pb(OAc)<sub>4</sub>], hydrolysis (OH<sup>-</sup>), and decarboxylation (100°C) (cf. reference 6).
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6. R. Young, and C. K. Chang, J. Am. Chem. Soc. **107**, 898 (1985).
7. 2-Trans: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 10.33 (s, 2H, meso), 5.23 (s, 2H, OH), 4.00 (m, 8H, CH<sub>2</sub>), 2.84 (m, 4H, CH<sub>2</sub>), 2.62 (m 4H, CH<sub>2</sub>), 1.87 (t, 12H, CH<sub>3</sub>), 0.96 (t, 12H, CH<sub>3</sub>) and naphthalene ring protons at 8.29 (d, 2H), 8.06 (d, 2H), 7.61 (d, 2H), 7.37 (t, 2H), 7.01 (t, 2H), and 6.84 (d, 2H); IR (KBr) 3530 cm<sup>-1</sup> (ν<sub>OH</sub>); mass spectrum m/e 818 (M<sup>+</sup>), 801 (M<sup>+</sup>-OH), 789 (M<sup>+</sup>-C<sub>2</sub>H<sub>5</sub>), 675 (M<sup>+</sup>-C<sub>10</sub>H<sub>7</sub>O), 662 (M<sup>+</sup>-C<sub>10</sub>H<sub>7</sub>O-CH<sub>2</sub>), 647 (M<sup>+</sup>-C<sub>10</sub>H<sub>7</sub>O-C<sub>2</sub>H<sub>5</sub>). The <sup>1</sup>H NMR, IR, and mass spectra of 2-cis were almost the same as those of 2-trans. Electronic spectra for CH<sub>2</sub>Cl<sub>2</sub> solutions: 2-trans, λ<sub>max</sub> (log ε) 413 (5.32), 510 (4.25), 545 (3.90), 576 (3.90), 628 nm (3.53); 2-cis, 413 (5.33), 510 (4.26), 545 (3.89), 576 (3.89), 628 nm (3.55).
8. 3: <sup>1</sup>H NMR for octamethylene moiety δ 3.20 (t, 4H, OCH<sub>2</sub>-), 0.43 (m, 4H, OCH<sub>2</sub>CH<sub>2</sub>-), -0.43 (m, 4H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), -1.51 (m, 4H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-); m/e 928 (M<sup>+</sup>).
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11. Eluent: n-hexane-2-propanol (99:1) (0.5 mL/min). Column dimension: 25 x 0.46 cm.

(Received in Japan 4 September 1986)